

# Antioxidant enzyme activities in strawberry fruit exposed to high carbon dioxide atmospheres during cold storage

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## Abstract

The effects of storage, in air or a 20% CO<sub>2</sub> in air (16.8% O<sub>2</sub>) atmosphere for 12 d at 2 °C, on antioxidant enzymes of strawberry fruit (*Fragaria x ananassa* Duch. cv. 'Jewel') were investigated. The concentrations of acetaldehyde, ethanol and ethyl acetate associated with fermentation were measured, and the activities of peroxidase (POX), catalase (CAT) and superoxide dismutase (SOD) assayed. Ethanol, acetaldehyde and ethyl acetate concentrations increased in CO<sub>2</sub>-treated fruit within a day of treatment, but more extensively after four days, while concentrations remained low in air-stored fruit. The total protein content extracted was not affected by CO<sub>2</sub> or storage time. Activities of POX were similar in air- and CO<sub>2</sub>-treated fruit, with an increase occurring only in air-treated fruit on day-12. Neither CAT nor SOD activities were affected by CO<sub>2</sub> treatment. In summary, a 20% CO<sub>2</sub> storage treatment induced fermentation but did not significantly affect total antioxidant enzyme activities.

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## 1. Introduction

Exposure of strawberries to controlled atmospheres (CA) with elevated CO<sub>2</sub> (12–20%) can extend storage periods of this perishable fruit by maintaining firmness and other quality attributes and reducing decay (Almenar, Hernández-Muñoz, Lagarón, Catalá, & Gavara, 2006; Harker, Elgar, Watkins, Jackson, & Hallett, 2000). CO<sub>2</sub> treatments of fruit in pallet shrouds are also used during strawberry transport (Mitcham & Mitchell, 2002). Strawberry metabolism is also strongly affected by CO<sub>2</sub> in the storage atmosphere (Fernández-Trujillo, Nock, & Watkins, 1999; Holcroft & Kader, 1999a, 1999b; Zhang & Watkins, 2005), and cultivars vary greatly in tolerances to CO<sub>2</sub> treatment (Fernández-Trujillo et al., 1999; Pelayo, Ebeler,

& Kader, 2003; Watkins, Manzano-Méndez, Nock, Zhang, & Maloney, 1999). One group, represented by 'Annapolis' and 'Cavendish', did not accumulate acetaldehyde and ethanol, while 'Honeye', 'Jewel', 'Kent' and especially 'Governor Simcoe', accumulated high levels of acetaldehyde and ethanol after 3–9 d when treated with 20% CO<sub>2</sub> (Watkins et al., 1999).

Accumulation of fermentation products in fruit exposed to elevated CO<sub>2</sub> indicates that these treatments represent significant stresses to the tissues. However, no information exists about the effects of CO<sub>2</sub> on the antioxidative defence systems in strawberry fruit. Peroxidase (POX) activity decreases, particularly in the earlier stages of maturity, during strawberry ripening (Civello, Martínez, Chaves, & Añón, 1995). Exogenous ethylene treatment in strawberry increases ionic leakage, fruit water loss and POX activity (Tian et al., 2000). Better quality of heat-treated strawberry fruit than untreated fruit was associated with lower POX activity, and higher ascorbic acid peroxidase (APX) and superoxide dismutase (SOD) activities, although differences occurred later in storage (Vicente, Martínez, Chaves, &

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Civello, 2006). Other antioxidant enzymes, e.g., ascorbate peroxidase, slightly increased during cold storage and particularly after a subsequent shelf-life of 1–2 d at 20 °C in strawberry (Vicente et al., 2006).

Research with other fruits suggests an involvement of antioxidative enzymes in postharvest stress responses. In pear fruit, SOD activity increased with higher CO<sub>2</sub> treatments during storage, while increased APX and decreased CAT activities were associated with development of brown core (Pintó, Lenthéric, Vendrell, & Larrigaudière, 2001). CA (2% O<sub>2</sub> and 5% CO<sub>2</sub>)-stored pears had a sharp burst of APX and glutathione reductase activities immediately after storage, and later increases and decreases of SOD and CAT activity, respectively (Larrigaudière, Lenthéric, Pintó, & Vendrell, 2001). In cherries, 12% CO<sub>2</sub> atmospheres show lower levels of POX activity during storage, irrespective of the O<sub>2</sub> concentration (Remón, Ferrer, López-Buesa, & Oria, 2004). In litchi, 70% O<sub>2</sub> treatments limited ethanol production of litchi flesh in the early period of storage and a slight decrease in POX activity was associated with pericarp browning (Tian, Li, & Xu, 2005). In peach, a storage treatment of 70% O<sub>2</sub>/0% CO<sub>2</sub> for 15 d, then changed to 5% O<sub>2</sub>/5% CO<sub>2</sub> at 0 °C, induced SOD and CAT activities (Wang, Tian, & Xu, 2005).

The objective of the current study was to investigate the activity of the antioxidative enzymes POX, CAT and SOD in response to elevated CO<sub>2</sub> treatment of 'Jewel' strawberry fruit in relation to accumulation of fermentation products.

## 2. Materials and methods

### 2.1. Plant material and experimental design

Fruit of the strawberry cultivar 'Jewel' were harvested at the Cornell Orchard (Ithaca, NY) at the white tip/orange stage of maturity. Fruit were sorted quickly to remove any that were damaged or under- or over-ripe. 20-fruit samples were placed into 1-l glass jars at 2 °C for 3 h to cool the fruit. Jars, with lids fitted with inlet and outlet ports, were then attached to a flow-through system (50 ml min<sup>-1</sup>) containing premixed atmospheres of air or 20% CO<sub>2</sub> in air. The gases were bubbled through water to humidify them, but the relative humidity was not measured. CO<sub>2</sub> and O<sub>2</sub> concentrations were measured by gas chromatography with a thermal conductivity detector equipped with a bridge current of 225 mA (Fisher Gas Partitioner, model 1200, Fisher Scientific, Springfield, NJ). The atmospheres at the inlet to jars were monitored at least once daily, and the outlet gas concentrations verified occasionally. The gas partitioner had a column temperature of 90 °C and injector temperature of 130 °C. Helium was used as the carrier gas with a flow rate of 30 ml min<sup>-1</sup>. There were two columns arranged in series; 80/100 mesh Porapak Q (6'6" × 1/8" i.d. SS, Supelco, Inc, Bellefonte, PA) and 60/80 mesh Molecular Sieve 13X (11' × 3/16" i.d. SS, Supelco, Inc, Bellefonte, PA). The partitioner output was recorded on a Hewlett/Packard 3394 integrator (Agilent, Wilming-

ton, DE). The gas chromatograph was calibrated with a standard mixture (4.97% CO<sub>2</sub> & 7.03% O<sub>2</sub> in a balance of N<sub>2</sub>; Airgas, Inc, Radnor, PA). To obtain 20% CO<sub>2</sub>, a constant stream of compressed CO<sub>2</sub> was added to a flow of compressed air through the use of needle valves. This resulted in an O<sub>2</sub> level of 16.8%. These concentrations were maintained within 1% of the desired levels throughout the experiment.

Jars with fruit were sampled in triplicate after the initial cooling period of 3 h, immediately after removal from storage, and after 1, 2, 4, 6, 8, 10 and 12 days. At each sampling time, calices were removed and approximately 100 g of berries were sliced immediately into liquid N<sub>2</sub>. The frozen fruit were kept at -80 °C prior to analysis.

### 2.2. Antioxidant enzyme activities

Strawberry tissue (25 ± 0.1 g), 50 ± 1 ml of buffer, was ground in a Waring blender with a semi-micro stainless steel container (increasing the speed two or three times for 20 s maximum). The buffer (modified from Rao, Watkins, Brown, & Weeden, 1998) contained 0.2 M potassium phosphate buffer, pH 7.8, 2% PVP-40, 2 mM EDTA, 50 mM CaCl<sub>2</sub>, and 1 mM phenylmethylsulfonyl fluoride (PMSF) dissolved in dimethyl sulfoxide (DMSO). The homogenate was transferred to a mortar kept in an ice bath for final grinding, then filtered through four layers of cheesecloth and collected. The pH was adjusted to 7.5 by adding HCl or NaOH solutions from 0.01 to 1 N while stirring in an ice bath, and then centrifuged at 25,000g for 30 min at 0–4 °C.

Acetone (-20 °C) was then added to the extracts in a ratio of 2:1 (v/w, ml acetone buffer per g of strawberry tissue). The extracts were kept at -20 °C for 10 min to precipitate proteins. Co-precipitation of pectin also occurred. The acetone extract was centrifuged at 10,000g for 10 min at 0–4 °C using Teflon tubes (Nalgene, Nalge Nunc International Corporation, Rochester, NY), and the supernatant discarded. The pellet was kept in an ice bath and dried using pure nitrogen flow. When no odour of acetone was detectable, the pellet was dissolved in a known volume of cold 0.1 M potassium phosphate buffer, pH 6.5 (usually 9 to 12 ml), using a vortex and an artist brush. This extract was stored in aliquots at -80 °C until used directly for enzymatic assays. Sometimes, Miracloth (Calbiochem, La Jolla, CA) was used to filter the final extracts and eliminate debris.

Protein content was determined according to Bradford (1976) using the standard assay of the BIORAD kit with BSA as a protein standard.

POX activity (EC 1.11.1.7) was assayed using guaiacol as the substrate (Rao et al., 1998). The assay mixture (3 ml) contained 100 mM phosphate buffer (pH 6.5), 13.3 mM guaiacol, and 500 µl of extract. A minimum of 25 µg of total protein was used per assay. The reaction was started by adding 10 µl of 30% H<sub>2</sub>O<sub>2</sub> and mixing gently three times in each cuvette. The changes of absorbance at

470 nm were recorded for 5 min at 10 s intervals, using a Beckman DU-60 spectrophotometer (Fullerton, CA). Activity was usually measured at 60 min.

CAT activity (EC 1.11.1.6) was assayed, according to Aebi (1984), by measuring the decomposition of  $H_2O_2$  at 240 nm. The assay mixture (1 ml in a quartz cuvette) contained 50 mM phosphate buffer (pH 7.0), and 3 mM  $H_2O_2$ , and 20  $\mu$ g protein (about 40–100  $\mu$ l of extract). Light yellow extracts are not useful for this assay, which requires really clear ones. Initial absorbance for the sample should be 0.5–0.9. CAT was usually calculated between 45 and 60 s.

SOD activity (EC 1.15.1.1) was assayed according to Asada, Takahashi, and Nagate (1974) and Rao, Paliyath, and Ormrod (1996) by measuring the reduction of ferricytochrome c. The following reagents, added sequentially, constituted the assay mixture of 3 ml: 50 mM sodium carbonate and sodium bicarbonate buffer (pH 10.2), 0.1 mM EDTA (pH 7.5), 0.1 mM xanthine, freshly prepared 0.015 mM ferricytochrome c (solution covered with aluminium foil) and 50  $\mu$ l of crude strawberry extract. The assay was performed simultaneously in triplicate, using a blank containing 50  $\mu$ l of 0.1 M potassium phosphate buffer at pH 7.5. The cuvettes were incubated at 25 °C for 3–4 min and the reaction initiated by adding 7.5–35  $\mu$ l of fresh xanthine oxidase solution (Sigma grade 1). The volume was adjusted to obtain a change in absorbance for the control from 0.032–0.034  $OD \cdot min^{-1}$ , but changes of 0.029–0.040  $OD \cdot min^{-1}$  were found during the life of the fresh solution. The changes in absorbance were monitored at 550 nm for 5 min. One SOD unit was defined as the amount of enzyme that inhibited the rate of ferricytochrome c reduction by 50% (Asada et al., 1974).

### 2.3. Volatile analysis

Acetaldehyde, ethanol and ethyl acetate were determined by headspace analysis, as previously reported (Fernández-Trujillo et al., 1999).

### 2.4. Statistical analysis

Data were subjected to ANOVA using general linear model procedures for calculation of least squared means and LSDs by using Minitab software v 11.12 (Minitab, Inc., State College, PA). Factors considered were storage time and atmosphere. Data from enzyme activities were transformed to natural logarithms for ANOVA.

## 3. Results

Concentrations of the fermentation products, acetaldehyde, ethanol and ethyl acetate, remained low in fruit stored in air, but increased over time in  $CO_2$ -treated fruit (Fig. 1). Accumulations of these compounds were generally greater after day-4.

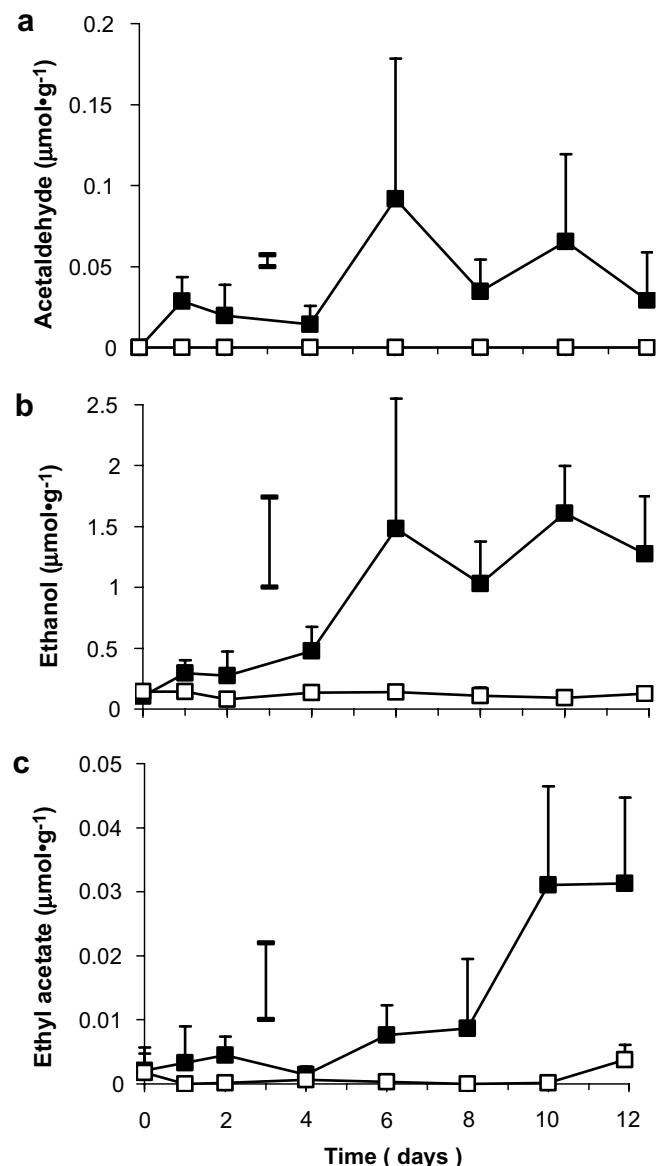


Fig. 1. Acetaldehyde (a), ethanol (b) and ethyl acetate (c) concentrations ( $\mu$ mol  $g^{-1}$  fresh weight) in strawberry cultivar 'Jewel' stored in air (open symbols) or 20%  $CO_2$ /16.8%  $O_2$  (closed symbols) at 2 °C for up to 12 d. Vertical bars represent LSD values at  $P=0.01$  for the interaction time  $\times$  atmosphere except in the case of acetaldehyde (LSD calculated for atmosphere).

Protein concentrations were not affected by atmosphere or storage time (data not shown), averaging  $34 \pm 10 \mu$ g/100  $\mu$ l extract.

POX activity in strawberry was affected by time, atmosphere and the time  $\times$  atmosphere interaction at  $P<0.05$ , but changes over time were inconsistent (Fig. 2a). However, activity of POX was much greater in air-stored fruit than in  $CO_2$ -treated fruit on day-12. CAT activity was not affected by  $CO_2$  treatment (Fig. 2b), but declined to a minimum after 8 d of storage, confirmed by polynomial trend analysis (cubic and fourth-order storage time effects were significant at  $P=0.002$  and  $P=0.04$ , respectively). No effects of time, atmosphere, or their interactions were

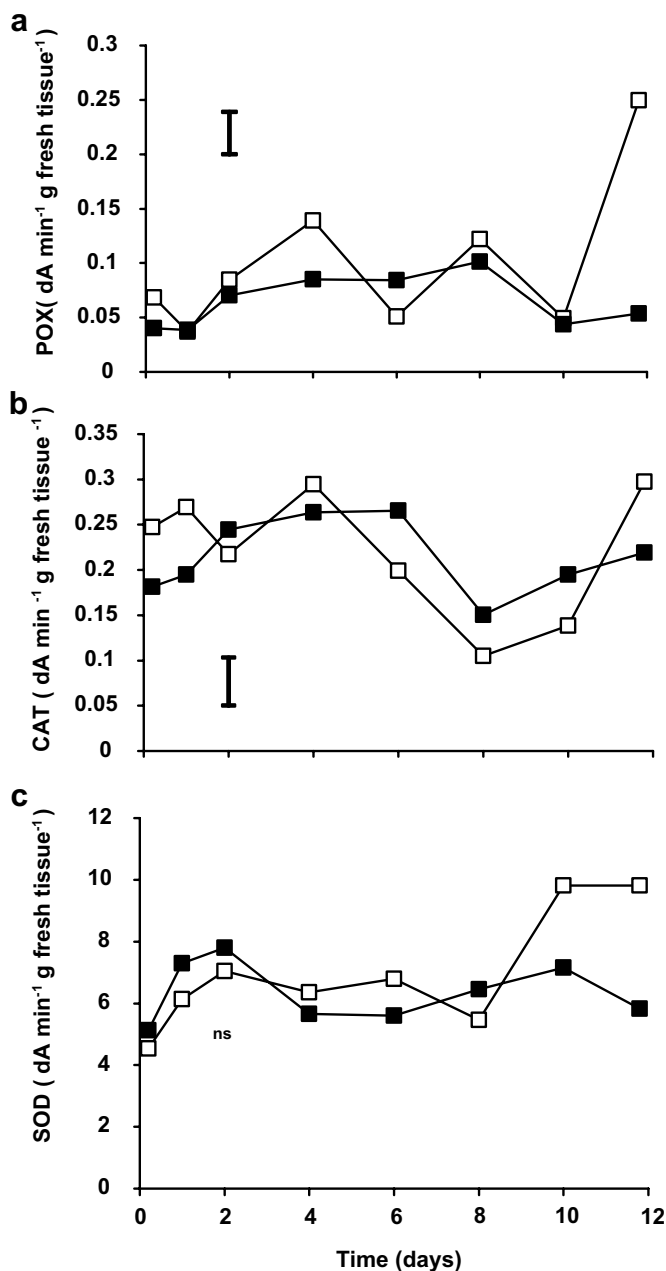


Fig. 2. Guaiacol-peroxidase (a), catalase (b), and superoxide dismutase (c) activities in fruit of strawberry cultivar 'Jewel' stored in air (open square) or 20% CO<sub>2</sub>/16.8% O<sub>2</sub> (closed square) at 2 °C for up to 12 d. Vertical bars represent LSD values at  $P = 0.01$  for the interaction time  $\times$  atmosphere (peroxidase activity) or the time effect (catalase activity). Ns means no significant effects.

detected on SOD activity (Fig. 2c), the average activity being  $6.7 \pm 2.4$  (SD) dA  $\cdot$  g fresh tissue<sup>-1</sup> min<sup>-1</sup> overall.

#### 4. Discussion

Accumulation of acetaldehyde, ethanol and ethyl acetate in response to CO<sub>2</sub> treatments is a common feature of strawberry fruit (Larsen & Watkins, 1995a, 1995b; Li & Kader, 1989), but the extent of accumulation is affected

by cultivar, ripening stage and the period of exposure to the gas (Fernández-Trujillo et al., 1999; Pelayo et al., 2003; Watkins et al., 1999). The 'Jewel' strawberries are moderately resistant to CO<sub>2</sub> (Fernández-Trujillo et al., 1999; Watkins et al., 1999), as shown by major accumulations occurring only after 4 d of treatment (Fig. 1).

However, with one exception, we were not able to detect differences in activities of POX, CAT or SOD activities between treatments. Higher POX activity was measured on day-12 in air-treated fruit than in CO<sub>2</sub>-treated fruit. This difference could be related to a delay in senescence by this treatment, as shown in apple fruit (Ingham, Parker, & Waldron, 1998). The possible role of isozymes that have been detected in strawberry (López-Serrano & Ros-Barceló, 1995) was not investigated in the current study, and such changes would not be detected by measurements of total activity. Civello et al. (1995) reported a decrease in POX activity during fruit ripening that was found primarily in a membrane-bound isoenzyme. Also, Holcroft et al. (1999a, 1999b) reported that inner and outer zones within the strawberry fruit exposed to high CO<sub>2</sub> may have different metabolisms that are not necessarily detectable by analyses of whole berries.

In conclusion, we have not been able to detect differences in antioxidant activities related to induction of fermentation in strawberry fruit due to high CO<sub>2</sub> atmospheres.

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